Glass Transition of Small Water Clusters

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Motivation

Amorphous ice

- the most abundant type of ice in the universe
- no molecular level studies
- small water clusters do not undergo crystallization

Main question

Do charged water clusters undergo a glass transition?

Sub-question

How many phase transitions do clusters undergo at temperatures $<100\ {\rm K}?$



Proposal

Goal

Observation of structural transition of small charged water clusters (H⁺(H₂O) or (H₂O)⁻ n = 20-200).



Means

RICE (Riken Cryogenic Electrostatic) storage ring

Methods

Photo-excitation and photo-fragmentation, mass and energy analysis of fragments, so called nano-calorimetry The first-ever storage ring studies of water clusters!

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What do we know so far



Phys. Rev. Lett., 103,

073401 (2009)

- No sharp transition
- Sometimes unphysical low heat capacity

- Transition temperature varies with size
- and converges to glass transition of LDA ice for bulk.

Multistep cooling model



Stillinger F. H., Science 267, 1935 (1995)

- ► Fast cooling rate → high final internal energy, broad distribution of structural states
- ► Slow cooling rate → low internal energy, narrow distribution of structural states
- ► $E_{\rm bar} \sim 100 \, {\rm meV}$ (Buck et al., PCCP 16, 6859, (2014))

We would prefer having control over cooling process and final structural states.

Nobody has! However, we can in Riken.

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Experiments in Riken



Nano-calorimetry



A skip in ΔE indicates the phase transition!

Other phase change indicators

- changes in vibrational spectra (hydrogen-bonded or free OH-stretching)
- changes in electron emission rate after laser irradiation (for (H₂O)⁻_n)

- Fragmentation dynamics
- Preferential fragmentation of isomers (pump-probe method)
- Annealing (irradiation by laser at the transition temperature)

etc.

Thank you for your attention

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Supplementary materials – potential profiles



Stillinger F. H., Science 267, 1935 (1995)

Supplementary materials – glass transition indications



Julien Boulon, Isabelle Braud, Sébastien Zamith, Pierre Labastie, and Jean-Marc L'Hermite JCP 140, 164305 (2014)

M. Schmidt and B. von Issendorff, JCP 136, 164307 (2012)





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Supplementary materials - rate of geometry change



Niedner-Schattenburg and Bondybey, Chem. Rev. 100, 11, 4059 (2000)

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